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Archaeometric study of roman pottery from Caudium area (Southern Italy)

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ABSTRACT - Aim of this work is the mineropetrographical characterization of late antique painted common wares from the ancient roman settlement of Caudium (today Montesarchio, Campania region, Italy).

Twenty-two samples (4th to 6th century AD) collected during the archaeological survey of the area, were studied to investigate their manufacturing technology and to attest a possible local production. Ceramics shards are represented by 16 painted common ware samples; furthermore, 2 bricks, 2 kiln rejects and 2 fragments of cooking ware were investigated for comparison.

Polarized light microscopy (PLM) observations and X-ray fluorescence chemical analyses (XRF) allowed to characterize the Caudium pottery production. Two main groups of fragments were distinguished: the first one composed by painted common wares, bricks and kiln rejects, the other one by cooking ware only. Two textural typologies were recorded within the first group (composed by calcareous clay pastes: average CaO ~ 11.0%), one characterized by fine pastes containing tiny clasts of quartz, feldspars and few volcanic inclusions, the other by coarser pastes with predominant volcanic temper. Cooking wares (composed by non-calcareous clay: average CaO ~ 1.6%) show a large amount of temper of both volcanic and detrital origin, the latter mainly constituted by

quartzarenite clasts. Multivariate statistical analysis (Hierarchical Clustering and Principal Component Analysis) confirms the already identified groups. Mineralogical analyses and scanning electron microscope observations of the sintering degree of clayey paste enabled to evaluate the firing temperatures of the most representative samples (from 800 to 1200°C).

The whole data set, along with geological features of the investigated area (wide availability of raw materials) and archaeological evidences (kiln refuses, large number of fragments of the same ceramic class), allowed to hypothesize a local production of the painted common wares and thus to define the respective reference group.

RIASSUNTO - Lo scopo di questo lavoro è la caratterizzazione minero-petrografica di frammenti di ceramica comune dipinta d'età tardo antica provenienti dall'antico insediamento romano di Caudium (attuale Montesarchio, Campania, Italia). Ventidue campioni (datati dal IV al VI secolo d.C.), ritrovati durante la ricognizione archeologica della zona, sono stati analizzati per ricostruire le tecnologie produttive e definire una possibile produzione locale. I reperti ceramici sono rappresentati da 16 frammenti di ceramica comune dipinta; inoltre sono stati investigati per confronto 2 laterizi, 2 scarti di fornace

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e 2 frammenti di ceramica da cucina. Al fine di caratterizzare i vari reperti sono state eseguite osservazioni in microscopia ottica a luce polarizzata e analisi chimiche in fluorescenza per raggi X. In questo modo è stato possibile individuare due raggruppamenti principali: il primo composto dalla ceramica comune dipinta, dai laterizi e dagli scarti di fornace, l'altro esclusivamente dalla ceramica da cucina. Fra i campioni del primo gruppo (composti da argilla calcarea: media CaO~11,0%) è possibile distinguere due tipologie tessiturali, una caratterizzata da impasti fini contenenti piccoli clasti di quarzo, feldspati e poche inclusioni vulcaniche, l'altra da impasti grossolani costituiti prevalentemente da degrassante vulcanico.

I frammenti di ceramica da cucina (composti da impasti non-calcarei: media CaO~1,6%) sono costituiti da abbondante degrassante, sia vulcanico che sedimentario, quest'ultimo formato prevalentemente da clasti quarzarenitici. L'analisi statistica multivariata ha confermato i raggruppamenti già individuati.

Tramite l'analisi mineralogica diffrattometrica su polveri e l'osservazione in microscopia elettronica del grado di sinterizzazione degli impasti sono state valutate anche le temperature di cottura di alcuni campioni rappresentativi (comprese fra 800 e 1200°C). Questi risultati, insieme con le caratteristiche geologiche (ampia disponibilità di materie prime) e archeologiche (presenza di scarti di fornace, gran numero di frammenti della stessa classe di ceramica) della zona in esame, hanno permesso di identificare una produzione locale limitata alla ceramica comune dipinta e di definirne il relativo gruppo di riferimento.

KEY WORDS: Caudium; reference group; painted common ware; Avellino Pumice; Ariano Unit.

INTRODUCTION

The archaeological area of Caudium is sited in a valley (*Valle Caudina*) located in the middle of the southern Apennine chain of Campania region (Italy). Caudium (today Montesarchio) can be considered the ancient capital of the *Caudini*, a Samnite tribe that peopled the area before the Roman conquest in the third century b.C. In the Roman period the city kept good trade exchanges especially with Naples and the Tyrrhenian area thanks to its favorable position along the *Via Appia*, the ancient roman road connecting Rome and *Brundisium* (Salmon, 1985). The occurrence of large clay outcrops and many kiln refuses in the same area, as well as a ceramic tradition so far preserved strongly suggest a local production of painted common ware.

The studied ceramic samples, the first investigated in the Caudium area, were all dated by comparison with some known shapes in late antique (4th to 6th century AD; Perrone, 2005).

The aim of this study is the archaeometrical characterization of a set of painted common ware fragments, in order to attest a local crafting of this typical ceramic production widespread in Southern Italy during Late Roman period. The archaeometrical data on this set of fragments were compared with a couple of kiln refuses attributed to the same painted common ware. Moreover two bricks and two cooking ware samples were also analyzed in order to verify differences or similarities of the technological features among local ceramic productions.

All the specimens come from three of the 151 sites forming the Caudium archaeological area (sites 111, 118, 121), discovered during an archaeological reconnaissance survey. This preliminary survey represents a very important step of an archaeological investigation necessary to reconstruct the ancient settlements arrangement, and to estimate a site frequentation on the basis of the surface distribution of the findings, especially ceramics shards.

GEOLOGICAL BACKGROUND

The Valle Caudina (Fig. 1) is a tectonic depression delimited by the Mesozoic carbonate massifs of the Taburno-Camposauro northwards, S. Agata de' Goti mountains westwards and Avella-Partenio chain southwards, the east side is characterized by the smoother morphologies of Miocene formations characterized by



Fig. 1 - Sketch map (1:250000) showing the main lithologies of the *Valle Caudina* area (modified after "Carta Geologica dell'Appennino Meridionale". Bonardi *et al.*, 1988).

silicoclastic sedimentary successions.

The field geology relieved some of the most representative Southern Apennine tectonostratigraphic units including the *Lagonegro Unit*, outcropping in the northwestern area of Montesarchio with the *Flysch Rosso* formation, a Late Cretaceous to Oligocene red marls and shales with interbedded coarse detrital limestone (Pescatore *et al.*, 1999).

In unconformity with the *Lagonegro Unit*, the *Altavilla Unit* (Messinian to Pliocene) crops out. Deposited in a *piggy-back* basin along the active border of the Apennine chain (Ori and Friend, 1984), this unit is formed by sandstones with silty clay layers; messinian evaporites are also present in the lower portion.

The succession continues with the Ariano Unit (Early to Middle Pliocene) represented by grayblue, well stratified marly and sandy clays. This unit was deposited in a *piggy-back* basin in open or shallow waters as showed by micropalaeontological evidence (De Castro Coppa *et al.*, 1969).

Silicoclastic turbiditic successions can be also recognized, such as the *Castelvetere flysch*, (Serravallian to Tortonian) formed by sandstones and conglomerates with plutonic-metamorphic and carbonate clasts (Critelli and Le Pera, 1995), also including mountain-block carbonate olistholiths of south Apennine carbonate platform edge (Ciarcia *et al.*, 2006).

Olistostromes of Argille Varicolori along with pelitic intercalations are also present in the *Fortore Unit* formation (Chiocchini, 2001), which outcrop principally on the Avella-Partenio slopes, not far from the southern side of the *Valle Caudina*.

In the Quaternary age the valley was covered by slope deposits, alluvial fans, fluvial-lacustrine sediments and volcanic products coming from Tyrrhenian margin eruptive vents: Campanian Ignimbrite (39 ka BP; De Vivo *et al.*, 2001; Fedele *et al.*, 2008) from Phlegraean Fields and Avellino Pumice of Somma-Vesuvius (4.3 ka BP; Santacroce *et al.*, 2008). Furthermore, the presence of other products of the Somma-Vesuvius activity cannot be excluded (e.g., AD 472 eruption).

MATERIALS AND METHODS

The 22 investigated shards (TABLE 1/A, 1/B), collected during the archaeological survey of the *Caudium* area, consist of 16 fragments of painted common ware (*PCW*) composed principally by basin rims (Fig. 2a, b) and amphora fragments (Fig. 2c), two shards of cooking ware (*CW*; Fig. 2d), two bricks (LAT 1, LAT 2) and two kiln rejects (SCF 1, SCF 2), these latter to be assumed as *PCW* wastes.

Macroscopic observations gave preliminary information about color, hardness, weight and fabric of the ceramic objects (Williams, 1990).

Polarized light microscopy (PLM) analysis of thin sections, using a "Leitz Laborlux 12 POL",

TABLE 1/A

Main macroscopic characteristics and petrographical features of ceramic fragments, the modal analysis data are shown as percentage in bold fonts.

Abbreviations (Siivola and Schmid, 2007): Qtz=Quartz, Afs=Alkalifeldspar, Cpx=Clinopyroxene, Ms=Muscovite, Bt=Biotite, Pl=Plagioclase, Am=Amphibole, Cal (sec)=Calcite (secondary), Lct=Leucite, Grt=Garnet, Mc=Microcline, Op=Opaque min., DRF=Detrital Rock Fragments, ARF=Argillaceous Rock Fragments, Tr.=Traces.

Porosity	medium- low	low	low	low	5.5%	medium low	low	very high	high	high	medium	very high	low	4.0%	medium	low	5.4%	1.9%	3.7%	4.4%	15.6%	10.7%
Grain sorting	seriate	seriate	seriate	seriate	bimodal	seriate	bimodal	bimodal	seriate	seriate	bimodal	seriate	bimodal	bimodal	seriate	bimodal	bimodal	bimodal	bimodal	bimodal	bimodal	bimodal
Packing	10-20 %	5-15 %	20-30 %	5-15 %	18.0%	10-20 %	15-25 %	5-15 %	5-15 %	10-20 %	5-15 %	10-20 %	10-20 %	23.1%	10-20 %	15-25 %	37.3%	36.4%	17.5%	31.2%	12.0%	17.8%
Optical properties	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic	isotropic
Weight (g)	22.9	18.5	38.0	32.7	35.0	17.2	9.6	17.0	30.5	94.7	19.5	11.3	79.5	74.1	43.3	86.3	34.6	17.4	13.9	16.5	42.4	20.1
Thickness (mm)	10	10	20	25	28	17	20	٢	10	20	7	6	13	40	10	15	45	15	18	5	45	20
Surface treatment	smoothed	coated	coated	smoothed, coated	smoothed, coated	,	coated	smoothed, coated		smoothed	smoothed, coated	smoothed, coated	smoothed, coated	smoothed, coated	smoothed	smoothed, coated	smoothed, coated	coated	·		coated	coated
Feel	rough	rough	rough	rough	rough	rough	rough	smooth	rough	rough	rough	rough	rough	rough	rough	rough	rough	rough	rough	rough	rough	rough
Hardness	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	hard	soft	hard	hard
Outer color	grayish green	light red	orange-red	red	grayish green	gray	light red	gray	gray	grayish ereen	grayish	gray	light red	yellow- brown	yellow	light red	dark red	light red	light gray	light red	dark gray	grayish green
Inner color	grayish green	dark red	orange-red	gray	gray ish green	gray	light red	gray	gray	grayish green	grayish green	gray ish green	dark red	red	greenish yellow	red	dark red	dark red	light gray	light red	grayish green	grayish green
Ceramic class	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	PCW	CW	CW	brick	brick	kiln waste	kiln waste
Sample	9-27	9-30	9-32	9-33	9-34	9-35	9-51	9-52	9-184	9-300	9-301	9-308	84-1	84-6	84-10	89-2	84-3	9-38	LAT 1	LAT 2	SCF 1	SCF 2

Sample	Qtz, Afs, DRF	Срх	Ms	Bt	Other minerals	Scoriae	Pumices	Volcanic lithics	ARF	Grog
9-27	XX	Х	Tr.	Tr.	Pl, Op	Х	Х	Х		
9-30	XX	Х	Х		Op	Х	Х	Х		
9-32	XXX	Tr.	Х	Х	Pl	Х	Х	Х	Х	Х
9-33	Х	Х	Х	Tr.		Х	Х	Х	Х	
9-34	12.1%	2.1%	0.8%	0.2%	Am (0.1%), Pl, Grt, Op (1.5%)	0.8%	0.3%	0.1%		
9-35	XX	Х	Tr.							
9-51	XXX	tr.	Х	Х	Pl, Op				Х	
9-52	Х	Tr.	Tr.	Tr.	Pl, Op			Х		
9-184	Х				Op					
9-300	Х				Op, Cal (sec)					
9-301	Х	Х	Tr.	Tr.	PI, Op, Cal (sec)	Х	Х	Х		
9-308	Х	Tr.	Х			Tr.				
84-1	XXX	Х	Х	Х	Pl	Х	Х	Х	Х	
84-6	16.3%	2.5%	0.8%	Tr.	Am (Tr.), Pl, Op (0.9%)	0.5%	1.3%	0.1%	0.7%	
84-10	XX	Х	Х	Tr.	Pl, Cal (sec), Op	Tr.			Х	
89-2	XX	Х	Tr.	Tr.	Pl, Op	Tr.	Х	Х	Х	
84-3	31.7%	1.0%		Tr.	Pl, Am (Tr.), Mc, Op (0.6%)	1.6%	2.0%	0.2%	0.2%	
9-38	34.0%	0.3%	Tr.	Tr.	Pl, Mc, Op (0.4%)	0.6%	0.6%	Tr.	0.5%	
LAT 1	10.6%	1.6%	0.3%	0.1%	Lct, Pl, Op (1.0%)	1.0%	1.5%	Tr.	1.4%	XX
LAT 2	11.8%	4.8%	1.5%	0.5%	Lct, Op (1.1%)	2.1%	4.5%	1.4%	3.5%	XX
SCF 1	0.4%	1.1%	1.5%	Tr.	Pl, Grt, Am (Tr.), Op (1.8%)	2.9%	3.6%	0.7%		
SCF 2	9.7%	1.1%			Pl, Op (2.0%)	1.3%	1.4%	2.3%		

TABLE 1/B

allowed to describe the texture, the matrix and the type of temper of the ceramic shards; a modal analysis (based on 4500 points) was also carried out (3.2.1 version of the "Leica Qwin" software equipped with a "Leica DFC 280" camera), to determine the relative abundance of inclusions,



Fig. 2 - Representative shapes of pottery finds from Caudium area. a) Sample 89-2. Painted common ware, basin rim fragment, 5th - 6th century AD (site 111). b) Sample 84-1. Painted common ware, basin rim fragment, 4th - 5th century AD (site 118). c) Sample 9-32. Painted common ware, amphora shards, 5th - 6th century AD (site 121). d) Sample 84-3. Cooking ware, pot fragment, 5th century AD (site 118).

matrix and pores.

Chemical analysis of major and trace elements was performed by X-ray fluorescence (XRF) with a "Philips PW1400" spectrometer (procedures and detection limits according to Melluso *et al.*, 2005). LOI (weight loss on ignition) was measured gravimetrically. Statistical treatment of chemical data (*Hierarchical Clustering and Principal Component Analysis*) was carried out with the 2.2.1 version of the "R Development Core Team" software.

X-ray powder diffraction (XRPD) was carried out with a "Philips PW 1730/3710" diffractometer (CuK α radiation 40 kV, 30 mA, curved graphite monochromator, scanning interval 3-80°, step size = 0.020° 2 θ , counting time 5s per step).

Scanning Electron Microscopy (SEM), performed with a "JEOL 5310" instrument, allowed to evaluate the sintering degree and the microtextural features of the clay matrix (Tite and Maniatis, 1975).

RESULTS

Macroscopic analysis

The majority of *PCW* samples show a green grayish colour; other samples of this class are red coloured with frequent zonings characterized by dark red or gray cores. A few fragments show red cores and creamy external surfaces. *CW* fragments are formed by red pastes with abundant temper grains. The brick LAT 1 shows a hard paste of light gray colour whereas LAT 2 sample is characterized by a red paste and a coarser texture. Hard gray greenish pastes typify kiln rejects with black edges in sample SCF 2.

Polarized light microscopy (PLM)

All analyzed samples have an isotropic matrix. PCW shards show two different textural characteristics. A group of fragments (84-10, 9-27, 9-30, 9-35, 9-52, 9-184, 9-300, 9-301, 9-308) show finer pastes with an average packing of about 10-15% and by seriate distribution (mean size range 15-150 µm) of the inclusions (Fig. 3a) constituted by tiny quartz grains, alkali feldspar and subordinated volcanics. The other group (84-1, 84-6, 89-2, 9-32, 9-33, 9-34, 9-51) is formed by samples showing a packing of about 15-25% and a bimodal distribution. The finer fraction (10-50 um) is constituted by quartz and alkali feldspar whereas the coarser one (100-200 μ m) mainly by volcanics (clinopyroxene, pumices, scoriae; Fig. 3b). Many samples of both groups also show minor mineral phases such as muscovite, biotite, plagioclase and opaque oxides. Amphibole and garnet only occur in samples 84-6 and 9-34. Two shards (84-1, 84-6) also contain few arenaceous rock fragments. Calcite was found as postdepositional pore-filling phases of samples 84-10, 9-300 and 9-301.

CW specimens are characterized by elongated fractures of the matrix and by a large amount of temper (packing 35-40%) with a bimodal grain



Fig. 3 - Thin sections images of some representative fragments. a) *PCW* sample 9-184. b) *PCW* sample 84-6. c) *CW* sample 9-38. d) *CW* sample 84-3. e) *CW* sample 84-3. f) Brick sample LAT 2. g) Brick sample LAT 1. h) Kiln reject sample SCF 1.

size distribution (Fig. 3c). Quartz and alkali feldspar occurs either in fine (10-50 μ m) or coarse fraction (100-200 μ m). The coarse fraction also shows fragments of arenitic rocks (Fig. 3d), quartz crystals (Fig. 3e) with undulating extinction, microcline, clinopyroxene, volcanic scoriae, pumices and ARF (*Argillaceous Rock Fragments*; Whitbread, 1986).

The bricks show a bimodal distribution of the grains and a packing of 15-20% in LAT 1 and of 30-35% in LAT 2. Quartz and alkali feldspar mainly occur in the fine fraction (10-50 μ m) whereas the coarse fraction (150-350 μ m) is composed by clinopyroxenes (diopside, Fesalite), leucite (Fig. 3f), pumices and scoriae of trachytic and leucititic composition. Leucite in scoriae shows faint polysynthetic twinning traces because of leucite-analcime transformation (Fig. 3g).

Kiln rejects are characterized by large rounded pores and dark green matrix in plane polarized light. Both samples present a bimodal distribution with a 10-20% packing with just few coarser grains. Alkali feldspar and quartz are mainly present in the fine fraction (10-50 μ m); coarser grains (150-250 μ m) are constituted by clinopyroxene, clinopyroxene-feldspar bearing lava fragments, scoriae, pumices and rare plagioclase crystals. Amphibole, biotite and garnet (Fig. 3h) are present in SCF 1 sample, while in SCF 2 analcimized leucite occurs in scoriae.

Modal analysis

Modal analysis (data shown in bold in TABLE 1/A and 1/B) was performed on eight representative samples to evaluate the relative abundances of grains, matrix and pores visible in thin section (Fig. 4). The two analyzed PCW samples (84-6, 9-34) show an average content of temper of 20.6%, mostly represented by a quartzfeldspathic detrital fraction (14.2%); the porosity is 4.7%. CW fragments (84-3, 9-38) are characterized by abundant temper (36.9%) with a quartz-feldspathic component of 32.8%; the porosity do not exceed 5.4%. Brick samples show a quite different content of temper (17.5% in LAT 1 and 31.2% in LAT 2). Both show an average porosity of about 4.0%. Temper in kiln rejects SCF 1 and SCF 2 is of 12.0% and 17.8%, respectively. Their mean porosity is 13.1%.

Chemical analysis (XRF)

As far as major and trace elements are



Fig. 4 - Relative abundances of inclusions, matrix and pores visible in some representative thin sections. *CW* sample 84-3, *CW* sample 9-38, *PCW* sample 84-6, *PCW* sample 9-34, Brick sample LAT1, Brick sample LAT2, Kiln reject sample SCF1, Kiln reject sample SCF2.

considered (TABLE 2), all PCW samples can be distinguished in two different groups. The CaO content ranging between 7.90 and 16.4% (average 10.9%), is always above 6% therefore proving the calcareous character of all samples attributed to this ceramic class (Maniatis and Tite, 1981). Only the sample 9-32, is slightly below this limit (CaO = 5.72%). Two groups can be distinguished: a first one (84-10, 9-27, 9-34, 9-35, 9-184, 9-300, 9-301, 9-308) characterized by higher values of CaO (on average 13.1%) and another one (84-1, 84-6, 89-2, 9-30, 9-32, 9-33, 9-51, 9-52) with a mean value of 8.62% (Fig. 5a). The first group also shows a lower average content of Al₂O₃ (14.2%) if compared with the remaining PCW specimens (16.2%) and a lower average values of Fe_2O_3 (5.65%) and TiO₂ (0.62%) if compared with the other samples of this ceramic class (Fe₂O₃ = 6.56%, TiO₂=0.73%; Fig. 5b).

The sample 9-52 shows highest K₂O (4.22%) and Na₂O (2.15%) values among all the *PCW* fragments, while the sample 9-51 reaches the highest P_2O_5 value (1.96%).

Concerning trace elements all the *PCW* samples show a quite homogeneous compositions making difficult to distinguish any grouping. To note the high Ba content (794 ppm) in sample 9-33.

As can be easily noticed in diagrams of figures 5a, b, c, the chemical composition of kiln rejects is similar to that of *PCW*, with the only exception of a slight higher Sr value (Fig. 5d).

Also the bricks have a chemical composition very similar to *PCW* samples; only strontium is on average higher in the bricks (439 ppm). Moreover the LAT 2 sample shows the highest Ba content (982 ppm).

CW samples can be distinguished from all the other analyzed fragments for the low CaO content (on average 1.54%). Other chemical features characterizing these two specimens are the average values of SiO₂ (68.0%) and MgO (1.99%), respectively the highest and the lowest among all studied samples. *CW* samples also show the lowest average contents of some trace elements (Fig. 5c, d), such as Zn (63 ppm), Ni (47 ppm) and Sr (196 ppm), whereas Zr (265 ppm) attains the highest value.

Statistical analysis

Statistical treatment of chemical data (*Hierarchical Clustering and Principal Component Analysis*) was performed depriving

| Ba | 422 | 458 | 461 | 516 | 343

 | 361

 | 406 | 404

 | 421.28 | 56.39 | | 588 | 568 | 587 | 505 | 562

 | 794

 | 611 | 325
 | 567.58 | 129.16 |
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 | 44 | 36
 | 11.48 | | 30 | 30 | 30
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| చ | 149 | 148 | 153 | 130 | 120

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 | 146 | 135

 | 137.22 | 13.94 | | 157 | 146 | 145 | 160 | 122

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 | 150 | 66
 | 140.86 | 20.33 |
 | 112
 | 84 | 98
 | 20.06 | | 146
 | 126 | 136
 | 14.38 | | 129 | 138 | 133
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| > | 134 | 143 | 145 | 129 | 117

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 | 143 | 130

 | 33.34 | 9.73 | | 138 | 135 | 131 | 146 | 124

 | 146

 | 125 | 119
 | 32.85 | 10.32 |
 | 138
 | 112 | 125
 | 17.95 | | 139
 | 138 | 139
 | 0.53 | | 140 | 143 | 141
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| Sc | 21 | 17 | 19 | 19 | 18

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 | 4.71 | | 25 | 22 | 23
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| Zr | 144 | 143 | 147 | 149 | 167

 | 162

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 | 0.43 1 | 8.81 | | 157 | 172 | 165 | 167 | 214

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 | 116

 | 138 | 122

 | 128.9 | 13.82 | | 142 | 132 | 135 | 138 | 163

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 | 13(| 18(
 | 144.49 | 17.7(|
 | 175
 | 22 | 198
 | 32.43 | | 168
 | 16 | 164
 | 4.8 | | 143 | 142 | 142
 |
| ž | 61 | 62 | 67 | 59 | 54

 | 56

 | 63 | 62

 | 60.58 | 4.23 | | 62 | 60 | 62 | 65 | 60

 | 67

 | 59 | 61
 | 62.16 | 2.60 |
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 | 44 | 47
 | 4.53 | | 68
 | 61 | 64
 | 5.03 | | 62 | 62 | 62
 |
| Zu | 66 | 103 | 88 | 93 | 16

 | 90

 | 98 | 95

 | 94.67 | 5.06 | | 114 | 103 | 116 | 122 | 132

 | 112

 | 112 | 93
 | 112.98 | 11.82 |
 | 99
 | 61 | 63
 | 3.65 | | 100
 | 103 | 101
 | 2.39 | | 101 | 67 | 66
 |
| FOI | 5.58 | 1.30 | 1.23 | 2.30 | 1.14

 | 1.48

 | 1.79 | 06.0

 | 1.96 | 1.52 | | 3.62 | 1.90 | 3.68 | 3.50 | 4.93

 | 4.66

 | 3.48 | 0.14
 | 3.24 | 1.55 |
 | 0.72
 | 0.96 | 0.84
 | 0.17 | | 2.00
 | 2.10 | 2.05
 | 0.07 | | 0.99 | 1.04 | 1.01
 |
| Sum | 100 | 100 | 100 | 100 | 100

 | 100

 | 100 | 100

 | | | | 100 | 100 | 100 | 100 | 100

 | 100

 | 100 | 100
 | | |
 | 100
 | 100 |
 | | | 100
 | 100 |
 | | | 100 | 100 |
 |
| P205 | 0.32 | 0.61 | 0.18 | 0.29 | 0.26

 | 0.32

 | 0.44 | 0.34

 | 0.35 | 0.13 | | 0.39 | 0.45 | 0.47 | 0.59 | 0.41

 | 0.61

 | 1.96 | 0.22
 | 0.64 | 0.55 |
 | 0.08
 | 0.01 | 0.04
 | 0.05 | | 0.39
 | 0.31 | 0.35
 | 0.05 | | 0.39 | 0.28 | 0.34
 |
| K20 | 2.51 | 2.77 | 2.75 | 2.90 | 2.70

 | 2.72

 | 2.75 | 2.77

 | 2.73 | 0.11 | | 2.99 | 2.86 | 2.97 | 2.97 | 3.33

 | 2.88

 | 3.00 | 4.22
 | 3.15 | 0.45 |
 | 2.99
 | 3.15 | 3.07
 | 0.12 | | 2.67
 | 3.16 | 2.92
 | 0.35 | | 3.34 | 2.96 | 3.15
 |
| Na ₂ O | 0.83 | 1.18 | 1.07 | 0.98 | 1.08

 | 1.20

 | 0.87 | 0.96

 | 1.02 | 0.13 | | 0.80 | 1.09 | 0.94 | 0.90 | 0.77

 | 0.83

 | 0.88 | 2.15
 | 1.05 | 0.46 |
 | 1.14
 | 1.27 | 1.21
 | 0.09 | | 1.73
 | 1.42 | 1.57
 | 0.22 | | 1.32 | 1.39 | 1.36
 |
| CaO | 16.43 | 12.18 | 13.55 | 11.96 | 12.48

 | 12.60

 | 13.26 | 12.37

 | 13.11 | 1.45 | | 7.90 | 10.14 | 7.94 | 9.31 | 5.72

 | 9.00

 | 8.70 | 10.25
 | 8.62 | 1.46 |
 | 1.86
 | 1.22 | 1.54
 | 0.45 | | 11.42
 | 8.99 | 10.20
 | 1.71 | | 11.49 | 11.82 | 11.66
 |
| MgO | 4.93 | 3.88 | 3.99 | 3.72 | 3.66

 | 3.76

 | 3.97 | 3.62

 | 3.94 | 0.42 | | 4.00 | 4.22 | 4.17 | 3.94 | 4.06

 | 4.23

 | 3.89 | 3.10
 | 3.95 | 0.37 |
 | 2.05
 | 1.94 | 1.99
 | 0.07 | | 4.59
 | 3.97 | 4.28
 | 0.44 | | 3.59 | 3.93 | 3.76
 |
| MnO | 0.10 | 0.12 | 0.11 | 0.11 | 0.14

 | 0.10

 | 0.13 | 0.14

 | 0.12 | 0.01 | | 0.14 | 0.12 | 0.13 | 0.14 | 0.16

 | 0.14

 | 0.17 | 0.15
 | 0.14 | 0.02 |
 | 0.14
 | 0.06 | 0.10
 | 0.06 | | 0.12
 | 0.17 | 0.15
 | 0.03 | | 0.12 | 0.12 | 0.12
 |
| Fe ₂ O ₃ | 5.46 | 5.86 | 5.75 | 5.75 | 5.53

 | 5.42

 | 5.63 | 5.84

 | 5.65 | 0.17 | | 6.99 | 6.33 | 6.83 | 6.75 | 6.83

 | 6.66

 | 6.63 | 5.46
 | 6.56 | 0.49 |
 | 6.92
 | 6.07 | 6.50
 | 0.60 | | 6.10
 | 6.52 | 6.31
 | 0.30 | | 5.74 | 5.78 | 5.76
 |
| N203 | 3.99 | 4.75 | 4.51 | 4.38 | 3.59

 | 3.58

 | 4.36 | 4.27

 | 4.18 | 0.42 | | 6.79 | 5.74 | 6.27 | 6.14 | 6.68

 | 5.93

 | 5.94 | 6.05
 | 6.19 | 0.37 |
 | 7.03
 | 6.63 | 6.83
 | 0.28 | | 5.58
 | 7.57 | 6.58
 | 1.41 | | 5.69 | 5.34 | 5.52
 |
| 102 | 19. | .64 | .63 | .63 | .60

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 | .74 | .59
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 | .70 | .74
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| SiO ₂ 1 | 4.81 (| 8.02 (| 7.47 0 | 9.28 (| 9.96 (

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 | 9.96 | 1.09 0 |
 | 7.01 0
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| | SiO ₂ TIO ₂ AL ₂ O ₃ Fe ₂ O ₃ MinO MgO CaO Na ₂ O K ₂ O P ₂ O ₃ Sum LOI Zin Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba | SiO ₂ 11O ₂ Al ₂ O ₃ Fe ₂ O ₃ MinO MgO CaO Na ₂ O K ₂ O P ₂ O ₅ Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba
54.81 0.61 13.99 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 134 149 30 422 | SiO ₂ 11O ₂ Al ₂ O ₃ Fe ₂ O ₃ MinO MgO CaO Na ₂ O K ₂ O P ₂ O ₅ Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba
54.81 0.61 13.99 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 134 149 30 422
58.02 0.64 14.75 5.86 0.12 3.88 12.18 1.18 2.77 0.61 100 1.30 103 62 150 401 33 143 14 17 143 148 22 458 | SiO2 1102 A1203 Fe203 Mi0 Mg0 Ca0 Naj0 K30 F305 Sim L01 Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba 54.81 0.61 13.99 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 134 149 30 422 58.02 0.64 14.75 5.86 0.12 3.88 12.18 1.18 2.77 0.61 100 1.30 62 150 401 33 143 14 17 143 148 22 458 57.47 0.63 14.51 5.76 0.11 3.99 13.55 1.07 2.75 0.18 100 123 88 67 135 390 34 147 18 195 145 153 20 461 | SiO2 TO2 TO2 FeO3 MiO WeO CaO NagO <td>SiO2 TO2 A12O3 Fe_AO3 MiO WegO CatO NagO <t< td=""><td>SiO2 TO2 A120 F≤O3 MO Wei Xi Xi <</td><td>SiO2 TO2 A:O3 F:O1 D:O3 SiO3 D:O3 <thd< td=""><td>SiO2 TO2 F4O3 F4O3 F4O3 F4O3 Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba< 54.81 0.61 139 5.46 0.10 4.33 6.31 0.32 100 5.58 99 61 113 385 31 144 21 21 149 30 422 458 58.02 0.64 14.75 5.86 0.11 3.90 156 147 18 147 18 149 30 422 448 57.47 0.63 14.51 5.75 0.11 3.70 12.06 133 149 17 18 147 18 148 22 446 59.28 0.63 14.51 5.75 0.11 3.70 12.06 13.26 0.18 100 129 159 159 153 20 416 15 55 14</td><td>SiO2 TO2 FeO3 Mod WeD Car Na Na <</td><td>SiO2 TO2 F4.03 F6.03 MO WO Cal Nage Col Nage Nage</td><td>SiO2 TO2 A:O3 F:O3 D:O3 Sum LO1 Zn< Ni< Rb Sr Y Zr< Nb Sc V C/2 Cu Ba< 54.81 0.61 139 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 148 23 20 461 57.47 0.63 14.51 5.75 0.11 3.99 13.55 107 2.75 0.18 100 1.30 13.5 149 14 18 14 17 148 23 20 461 57.47 0.61 1.35 1.01 2.75 0.18 100 1.30 133 149 18 17 16 343 353 30 149 17 148 17 149 17 149 30 19 36 36 36 36</td><td>SiO2 TO2 A_2O3 F_2O3 Mady Cad Mady Mady Cad Mady Cad Mady Mady Cad Mady Mady</td><td>SiO2 TO2 A40 F5O F3O F3O SiO3 SiO4 Dial Ni Ni</td><td>SiO2 TO2 A40 FeO MaD Mag Cad Pad V Zr Nb Sec V Cr Cuo Bad 54.81 0.61 139 5.46 0.10 4.39 16.43 0.83 251 0.32 103 153 135 113 385 31 144 21 21 149 30 44 57.47 0.63 1451 5.75 0.11 3.93 153 164 17 148 22 461 57.47 0.63 143 5.75 0.11 3.75 0.11 3.75 0.14 366 134 147 18 117 120 23 20 461 57.95 0.60 13.76 120 120 120 127 0.32 100 144 21 14 16 117 126 141 126 141 17 143 17 126 141 167 116<</td><td>SiO2 Holio Neio <t< td=""><td>SU01 TV1 Alia Ferol Mino Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Maid Mg0 Cal Naid Maid Mg0 Cal Maid Maid<td>SU0 TU0 Alio Fe(0) Mid Mid<</td><td>S00 100 Allob Fe(0) Math <th< td=""><td>NO. 110. A10. NMO NMO<!--</td--><td>SND HOD HED NED HED NED NED</td></td></th<><td>SU0 TU0 Tu0 Wi0 Ke0 Na0 Na0<td>SNO: TUO: <th< td=""><td>NO, HOO, HOO, HOO NOA NOA NO NOA NO NOA NO NOA NO NOA NOA</td><td>No. No. No.</td></th<><td>No. 100, FeeD, Mod Mod Kao Kao<</td><td>No. No. No.<td>No. No. No.<td>No. 10. A.10. 15.0. Mon Med Cad Name Cad Name Nam</td><td>No. No. No.<td>0 01 0300 010 0300<</td><td>No. No. No.<td>No. No. No.<td>No. No. No.<td>Sind Display Sind Display Sind Display Sind Display Display</td></td></td></td></td></td></td></td></td></td></td></t<></td></thd<></td></t<></td> | SiO2 TO2 A12O3 Fe_AO3 MiO WegO CatO NagO NagO <t< td=""><td>SiO2 TO2 A120 F≤O3 MO Wei Xi Xi <</td><td>SiO2 TO2 A:O3 F:O1 D:O3 SiO3 D:O3 <thd< td=""><td>SiO2 TO2 F4O3 F4O3 F4O3 F4O3 Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba< 54.81 0.61 139 5.46 0.10 4.33 6.31 0.32 100 5.58 99 61 113 385 31 144 21 21 149 30 422 458 58.02 0.64 14.75 5.86 0.11 3.90 156 147 18 147 18 149 30 422 448 57.47 0.63 14.51 5.75 0.11 3.70 12.06 133 149 17 18 147 18 148 22 446 59.28 0.63 14.51 5.75 0.11 3.70 12.06 13.26 0.18 100 129 159 159 153 20 416 15 55 14</td><td>SiO2 TO2 FeO3 Mod WeD Car Na Na <</td><td>SiO2 TO2 F4.03 F6.03 MO WO Cal Nage Col Nage Nage</td><td>SiO2 TO2 A:O3 F:O3 D:O3 Sum LO1 Zn< Ni< Rb Sr Y Zr< Nb Sc V C/2 Cu Ba< 54.81 0.61 139 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 148 23 20 461 57.47 0.63 14.51 5.75 0.11 3.99 13.55 107 2.75 0.18 100 1.30 13.5 149 14 18 14 17 148 23 20 461 57.47 0.61 1.35 1.01 2.75 0.18 100 1.30 133 149 18 17 16 343 353 30 149 17 148 17 149 17 149 30 19 36 36 36 36</td><td>SiO2 TO2 A_2O3 F_2O3 Mady Cad Mady Mady Cad Mady Cad Mady Mady Cad Mady Mady</td><td>SiO2 TO2 A40 F5O F3O F3O SiO3 SiO4 Dial Ni Ni</td><td>SiO2 TO2 A40 FeO MaD Mag Cad Pad V Zr Nb Sec V Cr Cuo Bad 54.81 0.61 139 5.46 0.10 4.39 16.43 0.83 251 0.32 103 153 135 113 385 31 144 21 21 149 30 44 57.47 0.63 1451 5.75 0.11 3.93 153 164 17 148 22 461 57.47 0.63 143 5.75 0.11 3.75 0.11 3.75 0.14 366 134 147 18 117 120 23 20 461 57.95 0.60 13.76 120 120 120 127 0.32 100 144 21 14 16 117 126 141 126 141 17 143 17 126 141 167 116<</td><td>SiO2 Holio Neio <t< td=""><td>SU01 TV1 Alia Ferol Mino Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Maid Mg0 Cal Naid Maid Mg0 Cal Maid Maid<td>SU0 TU0 Alio Fe(0) Mid Mid<</td><td>S00 100 Allob Fe(0) Math <th< td=""><td>NO. 110. 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Mon Med Cad Name Cad Name Nam</td><td>No. No. No.<td>0 01 0300 010 0300<</td><td>No. No. No.<td>No. No. No.<td>No. No. No.<td>Sind Display Sind Display Sind Display Sind Display Display</td></td></td></td></td></td></td></td></td></td></td></t<></td></thd<></td></t<> | SiO2 TO2 A120 F≤O3 MO Wei Xi Xi < | SiO2 TO2 A:O3 F:O1 D:O3 SiO3 D:O3 D:O3 <thd< td=""><td>SiO2 TO2 F4O3 F4O3 F4O3 F4O3 Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba< 54.81 0.61 139 5.46 0.10 4.33 6.31 0.32 100 5.58 99 61 113 385 31 144 21 21 149 30 422 458 58.02 0.64 14.75 5.86 0.11 3.90 156 147 18 147 18 149 30 422 448 57.47 0.63 14.51 5.75 0.11 3.70 12.06 133 149 17 18 147 18 148 22 446 59.28 0.63 14.51 5.75 0.11 3.70 12.06 13.26 0.18 100 129 159 159 153 20 416 15 55 14</td><td>SiO2 TO2 FeO3 Mod WeD Car Na Na <</td><td>SiO2 TO2 F4.03 F6.03 MO WO Cal Nage Col Nage Nage</td><td>SiO2 TO2 A:O3 F:O3 D:O3 Sum LO1 Zn< Ni< Rb Sr Y Zr< Nb Sc V C/2 Cu Ba< 54.81 0.61 139 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 148 23 20 461 57.47 0.63 14.51 5.75 0.11 3.99 13.55 107 2.75 0.18 100 1.30 13.5 149 14 18 14 17 148 23 20 461 57.47 0.61 1.35 1.01 2.75 0.18 100 1.30 133 149 18 17 16 343 353 30 149 17 148 17 149 17 149 30 19 36 36 36 36</td><td>SiO2 TO2 A_2O3 F_2O3 Mady Cad Mady Mady Cad Mady Cad Mady Mady Cad Mady Mady</td><td>SiO2 TO2 A40 F5O F3O F3O SiO3 SiO4 Dial Ni Ni</td><td>SiO2 TO2 A40 FeO MaD Mag Cad Pad V Zr Nb Sec V Cr Cuo Bad 54.81 0.61 139 5.46 0.10 4.39 16.43 0.83 251 0.32 103 153 135 113 385 31 144 21 21 149 30 44 57.47 0.63 1451 5.75 0.11 3.93 153 164 17 148 22 461 57.47 0.63 143 5.75 0.11 3.75 0.11 3.75 0.14 366 134 147 18 117 120 23 20 461 57.95 0.60 13.76 120 120 120 127 0.32 100 144 21 14 16 117 126 141 126 141 17 143 17 126 141 167 116<</td><td>SiO2 Holio Neio <t< td=""><td>SU01 TV1 Alia Ferol Mino Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Maid Mg0 Cal Naid Maid Mg0 Cal Maid Maid<td>SU0 TU0 Alio Fe(0) Mid Mid<</td><td>S00 100 Allob Fe(0) Math <th< td=""><td>NO. 110. 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Mon Med Cad Name Cad Name Nam</td><td>No. No. No.<td>0 01 0300 010 0300<</td><td>No. No. No.<td>No. No. No.<td>No. No. No.<td>Sind Display Sind Display Sind Display Sind Display Display</td></td></td></td></td></td></td></td></td></td></td></t<></td></thd<> | SiO2 TO2 F4O3 F4O3 F4O3 F4O3 Sum LOI Zn Ni Rb Sr Y Zr Nb Sc V Cr Cu Ba< 54.81 0.61 139 5.46 0.10 4.33 6.31 0.32 100 5.58 99 61 113 385 31 144 21 21 149 30 422 458 58.02 0.64 14.75 5.86 0.11 3.90 156 147 18 147 18 149 30 422 448 57.47 0.63 14.51 5.75 0.11 3.70 12.06 133 149 17 18 147 18 148 22 446 59.28 0.63 14.51 5.75 0.11 3.70 12.06 13.26 0.18 100 129 159 159 153 20 416 15 55 14 | SiO2 TO2 FeO3 Mod WeD Car Na Na < | SiO2 TO2 F4.03 F6.03 MO WO Cal Nage Col Nage Nage | SiO2 TO2 A:O3 F:O3 D:O3 Sum LO1 Zn< Ni< Rb Sr Y Zr< Nb Sc V C/2 Cu Ba< 54.81 0.61 139 5.46 0.10 4.93 16.43 0.83 2.51 0.32 100 5.58 99 61 113 385 31 144 21 21 148 23 20 461 57.47 0.63 14.51 5.75 0.11 3.99 13.55 107 2.75 0.18 100 1.30 13.5 149 14 18 14 17 148 23 20 461 57.47 0.61 1.35 1.01 2.75 0.18 100 1.30 133 149 18 17 16 343 353 30 149 17 148 17 149 17 149 30 19 36 36 36 36 | SiO2 TO2 A_2O3 F_2O3 Mady Cad Mady Mady Cad Mady Cad Mady Mady Cad Mady Mady | SiO2 TO2 A40 F5O F3O F3O SiO3 SiO4 Dial Ni Ni | SiO2 TO2 A40 FeO MaD Mag Cad Pad V Zr Nb Sec V Cr Cuo Bad 54.81 0.61 139 5.46 0.10 4.39 16.43 0.83 251 0.32 103 153 135 113 385 31 144 21 21 149 30 44 57.47 0.63 1451 5.75 0.11 3.93 153 164 17 148 22 461 57.47 0.63 143 5.75 0.11 3.75 0.11 3.75 0.14 366 134 147 18 117 120 23 20 461 57.95 0.60 13.76 120 120 120 127 0.32 100 144 21 14 16 117 126 141 126 141 17 143 17 126 141 167 116< | SiO2 Holio Neio Neio <t< td=""><td>SU01 TV1 Alia Ferol Mino Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Ferol Naid Mg0 Cal Naid Maid Mg0 Cal Naid Maid Mg0 Cal Maid Maid<td>SU0 TU0 Alio Fe(0) Mid Mid<</td><td>S00 100 Allob Fe(0) Math <th< td=""><td>NO. 110. 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TABLE 2



Fig. 5 - XRF binary diagrams of major (wt.%) and traces elements (ppm). Dots indicate *PCW*, triangles indicate *CW*, rhombs indicate bricks, squares indicate kiln rejects.

starting data set of some chemical components, in particular CaO, MnO, P_2O_5 and Ba, as they could have been affected by post burial contamination processes (Fabbri *et al.*, 1994; Maggetti, 2001). Afterwards, data were log_{10} transformed in order to avoid deleterious effects of scale effects of clustering results (e.g., V-shaped chemical data; Aruga, 2003; Hall, 2004). In order to further simplify statistical procedures Principal Components Analysis (PCA) was carried out.

The cumulative proportion of variance, which explained the 96% of the problem, is at the 7th component (TABLE 3). Thus, the seven variables that mostly affect the problem were chosen (Hall, 2004): Fe₂O₃, MgO, Na₂O, Y, Sc, V and Cu. The dendrogram (Fig. 6) resulting from the Hierarchical Clustering Analyses (*average* method, *euclidean* distance) enabled to distinguish two groups. The first one (group 1) is further subdivided in three subgroups: 1a, including *PCW* and kiln rejects; 1b, formed by bricks; 1c, formed by the three samples of *PCW* which showed the lowest values of Y. The group 2 is characterized by the *CW* samples and a *PCW* fragment (84-3).

Mineralogical analysis (XRPD)

X-ray diffraction data (TABLE 4) confirmed the prevailing quartz-feldspathic composition of all samples as evidenced by PLM. Quartz prevails in CW(84-3, 9-38) while is less abundant in the kiln rejects (SCF1, SCF2). Feldspar was recorded in trace in the 9-52 PCW sample. Calcite only occurs in 84-10; illite was clearly observed in 9-32 and is only in trace in 84-1, 9-30, 9-51 and 9-33. Pyroxene is ubiquitous except for samples 84-1, 84-3, 89-2, 9-32, 9-33, 9-38. A low amount of gehlenite was observed in 84-1, 9-30, 9-33, 9-51, LAT 1, whereas occurs only in traces in samples 84-6, 84-10, 89-2, and 9-27. Leucite was found in the two bricks (LAT 1, LAT 2) and in the kiln refuses (SCF 1, SCF 2). Analcime was recorded in LAT 1, SCF 2 and only in trace in LAT 2.

Scanning electron microscopy (SEM)

The observation of ceramic paste by scanning electron microscope (SEM) provided information

TABLE 3

Principal component analysis. The table shows the weight of the variables of each principal component. Standard deviation (σ), variance and cumulative proportion are also reported.

Component	1	2	3	4	5	6	7
SiO ₂	-0.303	-	0.290	-0.108	0.104	-	-
TiO_2	-	0.487	0.200	-	-	-	-
Al_2O_3	-0.172	0.429	-0.171	-	-	-	-
Fe ₂ O ₃	-	0.497	0.157	-	-	-	-0.127
MgO	0.325	-	-0.107	-	-0.211	-	-0.317
Na ₂ O	-0.128	-0.181	-0.498	0.185	0.246	0.141	-0.168
K ₂ O	-0.192	0.106	-0.412	-	-0.407	-	0.497
Zn	0.240	0.255	-0.102	-	-0.487	-	-
Ni	0.287	0.123	-0.249	0.128	-	-0.320	-
Rb	-0.291	0.117	-0.23	-	0.104	-0.325	-0.314
Sr	0.295	-	-0.307	-	0.104	0.158	-0.117
Y	-	-	0.152	0.768	-	0.150	0.267
Zr	-0.320	0.127	-	0.167	-	0.214	-
Nb	-0.183	0.176	-0.36	-0.397	0.155		0.122
Sc	0.292	-	-	-0.312	-	0.143	0.518
V	0.210	0.192	-	0.154	0.544	-0.378	0.340
Cr	0.311	0.180	-	-	-	-0.232	-
Cu	0.191	0.256	-	-	0.349	0.676	-
σ	2.82	1.90	1.47	1.21	1.01	0.84	0.64
Variance	0.44	0.20	0.12	0.08	0.06	0.04	0.02
Cumulative proportion	0.44	0.64	0.76	0.85	0.90	0.94	0.96

on the sintering degree and the pore structure developed in the clay matrix after the firing process. This allowed to evaluate the firing temperatures as suggested by the thermal stability of the mineral phases detected by XRPD analysis (Tite and Maniatis, 1975; Maniatis and Tite, 1981). The estimation of the firing temperatures was carried out on 11 representative samples chosen on the basis of their mineralogical composition. TABLE 5 reports some technological features of the samples (firing atmosphere and firing temperatures) hypothesized according to the vitrification stages of the clay matrix (Fig. 7).

DISCUSSION

A group of *PCW* samples (84-10, 9-27, 9-34, 9-35, 9-52, 9-184, 9-300, 9-301, 9-308) show very hard pastes of green grayish colour, likely due to high firing temperatures of calcium-rich clays (Picon, 2002), as confirmed by XRPD and SEM analyses. Hard and coarser red coloured pastes, sometimes showing sandwich zoning with dark red or gray cores, characterize the other samples of this ceramic class. Some specimens (84-6, 84-10) show red or yellow-red cores and creamy edges. This shading red hues of the outer layers of the ceramic walls is the



Fig. 6 - Hierarchical clustering dendrogram representation (average method, euclidean distance).

IABLE 4
Semiquantitative mineralogical evaluation (XRPD) of investigated samples.
xxxx = predominant, xxx = abundant, xx = frequent, x = sporadic, tr = traces

4

Sample	Quartz	Feldspar	Calcite	Hematite	Illite	Gehlenite	Clinopyroxene	Analcime	Leucite
9-27	XX	XXX	-	-	-	traces	XX	-	-
9-30	XX	XXX	-	traces	traces	Х	traces	-	-
9-32	XXX	XXX	-	traces	Х	-	-	-	-
9-33	XXX	XXX	traces	Х	traces	Х	-	-	-
9-34	XX	Х	-	XX	-	-	XX	-	-
9-35	XXX	Х	-	XX	-	-	XX	-	-
9-51	XXX	XX	-	Х	traces	Х	traces	-	-
9-52	XXX	traces	-	XX	-	-	XX	-	-
9-184	XX	Х	traces	-	-	-	XX	-	-
9-300	XX	XX	-	XX	-	-	XX	-	-
9-301	XX	Х	-	XX	-	-	XX	-	-
9-308	XX	Х	-	XX	-	-	XX	-	-
84-1	XXX	XXX	-	Х	traces	Х	-	-	-
84-6	XX	XXX	-	Х	-	traces	XX	-	-
84-10	XXX	XX	XX	-	-	traces	XX	-	-
89-2	XX	XXX	-	traces	-	traces	-	-	-
84-3	XXXX	XX	-	Х	-	-	-	-	-
9-38	XXXX	XX	-	traces	-	-	-	-	-
LAT 1	XX	XXX	-	traces	-	Х	XX	Х	Х
LAT 2	XX	XXX	-	Х	-	traces	XX	traces	Х
SCF 1	Х	XXX	-	Х	-	-	XX	-	Х
SCF 2	Х	XX	-	Х	-	-	XX	Х	Х

85

TABLE 5

Firing temperature ranges of some representative samples. Type of clay: NC = Non calcareous, C = Calcareous; firing atmosphere: ox = oxidizing, red = reducing; vitrification stages: IV = initial vitrification, V =extensive vitrification, CV = continuous vitrification, (MB) = medium bloating pores, (CB) = coarse bloating pores; abbreviations according to Tite and Maniatis, (1975).

Sample	Type of clay	Firing atmosphere	Vitrification stage	Firing temperatures (°C)
9-32	NC	ox	IV	800 - 850
84-1	С	ox	V	850 - 950
84-6	С	ox	V	850 - 1050
9-33	С	ox	V	850 - 950
LAT 1	С	ox	V	850 - 1050
LAT 2	С	ox	V	850 - 1050
84-3	NC	ox	CV	950 - 1000
84-10	С	ox	CV(MB)	1050 - 1200
9-52	С	ox	CV(CB)	1050 - 1200
SCF 1	С	ox - red	CV(CB)	1050 - 1200
SCF 2	С	ox - red	CV(CB)	1050 - 1200



Fig. 7 - SEM images of ceramic pastes. a) *PCW* sample 9-32, vitrification stage IV; b) *PCW* sample 84-1, vitrification stage V. c) *CW* sample 84-3, vitrification stage CV. d) Kiln reject sample SCF 2, vitrification stage CV(CB). Abbreviations according to Tite and Maniatis (1975).

results of the Fe³⁺ incorporation in the structures of new forming calcium silicates (e.g., clinopyroxene as revealed by XRPD) in the calcium-rich pastes fired at high temperatures (Molera *et al.*, 1998).

The optically isotropic matrix observed in optical microscopy may also accounts for the quite high firing temperatures of all the investigated samples. Most of the PCW fragments (84-10, 9-27, 9-30, 9-35, 9-52, 9-184, 9-300, 9-301, 9-308) are characterized by fine pastes with serial distribution of grains mainly represented by quartz, feldspar and mica with few volcanics. The other PCW samples (84-1, 84-6, 89-2, 9-32, 9-33, 9-34, 9-51) show a bimodal distribution with a relatively abundant coarse fraction mostly constituted by volcanic inclusions (clinopyroxene, pumices, lithics, scoriae). PCW samples display a chemical composition reflecting in many cases the textural grouping observed in PLM. A group formed by all the finer pastes PCW samples (84-10, 9-27, 9-34, 9-35, 9-184, 9-300, 9-301, 9-308) but also including the fragment 9-34 of coarser texture, is characterized by higher values of CaO and lower contents of Al₂O₃, Fe₂O₃ and TiO₂ in comparison with the group of PCW fragments (84-1, 84-6, 89-2, 9-30, 9-32, 9-33, 9-51, 9-52), mainly composed by coarser paste samples (the finer ones being 9-30 and 9-52). The anomalous P_2O_5 content (1.96%) of sample 9-51 may be due to post-burial contamination phenomena (Fabbri et al., 1994; Maggetti, 2001).

CW specimens also showing optically isotropic matrix, are characterized by abundant temper mainly constituted by detrital quartzarenitic rock fragments, which account for the higher SiO₂ values observed (68 wt%); some metamorphic rock fragments and frequent volcanics are present as well. They are marked by a strong non-calcareous character (average CaO = 1.54%) due to the use of a different clay raw material.

The bricks are characterized by a chemical

composition similar to that of *PCW* samples. The highest values of Al_2O_3 (17.6%) and some trace elements (Ba, Sr) recorded in LAT 2 are likely due to the larger abundance of temper (31.2%) if compared to that of LAT 1 (17.5%) and all *PCW* samples. Among the volcanic products observed in these specimens leucite stands out either as individual crystals or analcimized in scoriae.

Kiln rejects, as expected, show marked evidences of high firing temperatures, as testified by their very hard pastes and by the large rounded pores as evidenced by SEM observations. Also, the dark gray colour is another effect of high temperatures. In particular, the dark green core fading in black edges of SCF 2 could be due to reducing conditions of the furnace atmosphere as a result of oxygen deficiency caused by the combustion of large quantities of wood (Picon, 2002). Kiln refuses show a chemical-petrographical composition very similar to that of PCW, with the only exception of higher Sr value. The occurrence of amphibole, garnet (SCF 1), and scoriae with analcimized leucite crystals typical of high potassium silica undersaturated rocks (HKS; Conticelli and Peccerillo, 1992) allows to correlate them to the Somma-Vesuvius volcanic complex, whose products are widespread in the valley (Zanchetta et al., 2004).

The statistical treatment of chemical data confirms the existence of a large group of PCW samples including the kiln rejects and the bricks; the two CW fragments are well separated from the other specimens.

The evaluation of the firing temperatures according to Tite and Maniatis (1975) points out for most samples temperatures above 850° C. Only for one *PCW* sample (9-32), the presence of illite and the initial vitrification stage (IV), indicate a firing temperature ranging between 800 and 850° C.

Five samples (84-1, 84-6, 9-33, LAT 1, LAT 2) show an extensive vitrification stage (V),

typical of firing temperatures between 850 and 1050°C. The presence of traces of gehlenite, starting to form at around 800°C (Cultrone et al., 2001) along with the persistence of illite, generally destroyed at around 950°C (Jordán et al., 1999) leads to hypothesize a range of 850-950°C for the 84-1 sample. The same considerations can be done also for the PCW sample 9-33. A firing temperature of 950-1000°C was assumed for the non calcareous CW sample 84-3 due to a continuous vitrification stage (CV). The persistence of gehlenite in traces along with neoformed clinopyroxene and a continuous vitrification stage with medium bloating pores CV(MB) in the PCW sample 84-10 account for temperatures next to 1050°C.

The continuous vitrification stage with coarse bloating pores CV(CB) observed in the kiln rejects (SCF 1, SCF 2) points out temperatures at least higher than 1050°C. The same firing temperatures were evaluated for the 9-52 *PCW* sample where a CV(CB) vitrification stage and the presence of neoformed diopside were recorded. The occurrence of analcime in the bricks (LAT 1, LAT 2) and in the kiln rejects (SCF 1, SCF 2) is likely do to by leucite alteration or even related to the alteration of the glassy phase (Buxeda I Garrigós *et al.*, 2002); nevertheless, analcime cannot be formed during the firing of the ceramics because of its high content of crystal water (Schwedt *et al.*, 2006).

CONCLUSIONS

The specimens of the present research represent the first ceramic findings of the Caudium area carefully analyzed. This site is particularly important for its strategic position along the *Via Appia*, the consular road connecting the west and the east coast of Italic Paeninsula since Roman period.

Furthermore, the discovery of many late antique findings witnesses the intense occupation of the *Valle Caudina* despite the scarce documentation available for that period.

This archaeometric investigation was focused on a large set of ceramic fragments, namely painted common ware, widespread in Southern Italy during Late Roman and Byzantine period (Grifa *et al.*, 2008), discovered during the archaeological reconnaissance survey of the site. Other samples of different type and function (cooking ware, bricks and kiln rejects) were also studied as comparison.

Painted common ware (*PCW*), bricks and kiln rejects are all characterized by calcareous pastes, presumably crafted with the same clayey raw material, mixed with different proportions of temper and fired between 850 and 1200°C.

CW samples show different chemical (e.g. low CaO) and petrographical features if compared to the *PCW* group so that the use of a different clayey raw material and temper is hypothesized. A possible exploitation area of the raw material should be one of the several non calcareous clay outcrops (*Argille Varicolori*) falling within the surrounding area of Caudium. Firing temperatures of about 950-1000°C were estimated for these samples.

The whole dataset, along with the archaeological (large number of fragments in the same site and the presence of kiln rejects) and geological (outcrop referred to blue-gray common clays of Ariano Unit, extensively used for ceramic production; Grifa *et al.*, 2009) evidences indicate a local production of painted common ware (*PCW*) and allow to identify its reference group (Maggetti, 2001).

The volcanic temper observed in the samples has been ascribed to the Avellino Pumice, a Somma-Vesuvius related deposit, widespread in the *Valle Caudina*, whose products were used for other pottery productions of the neighbour Benevento area (e.g., Protomajolica; Grifa *et al.*, 2006, 2009). Nevertheless, leucite found in kiln rejects and bricks, is a mineral whose occurrence is not attested in this deposit and has to be related to other Somma-Vesuvius products. It would be possibly associated to the Pollena Eruption (AD 472), a leucite-bearing product, whose dispersal area overlaps the *Valle Caudina* (Zanchetta *et al.*, 2004). In this case, the use of volcanic temper related to Pollena Eruption could likely indicate the AD 472 as the lower limit of manufacturing for bricks and kiln rejects, these latter closely correlated with painted common ware (*PCW*).

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